

Supporting Information

Recovery of Ammonium and Phosphate using Battery Deionization in a Background Electrolyte

Moon Son ^a, Benjamin L. Aronson ^a, Wulin Yang ^a, Christopher A. Gorski ^a, Bruce E. Logan ^{a,*}

^a Department of Civil and Environmental Engineering, The Pennsylvania State University,
University Park, PA 16802, USA

* Corresponding author. Email: blogan@psu.edu; Tel.: +1-814-863-7908

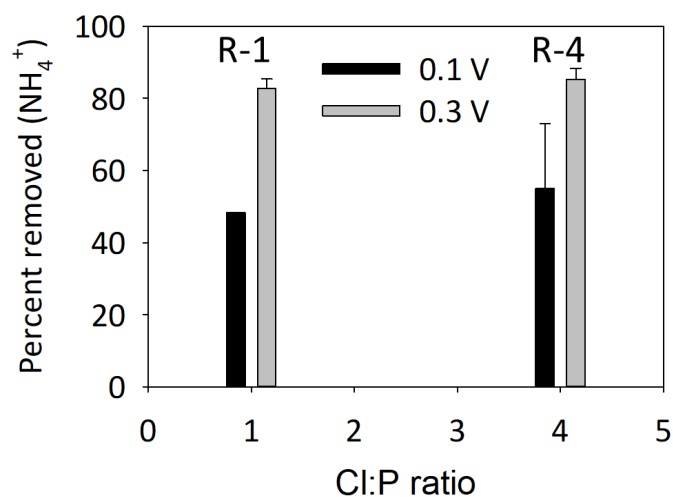


Fig. S1 Removal of NH_4^+ at the applied constant voltages of 0.3 or 0.1 V as a function of chloride/phosphorus (Cl:P) ratio. NH_4^+ (5 mM) concentrations remained constant and the R1 (Cl:P ration of 1) and R-4 (Cl:P ration of 1) solutions were used.

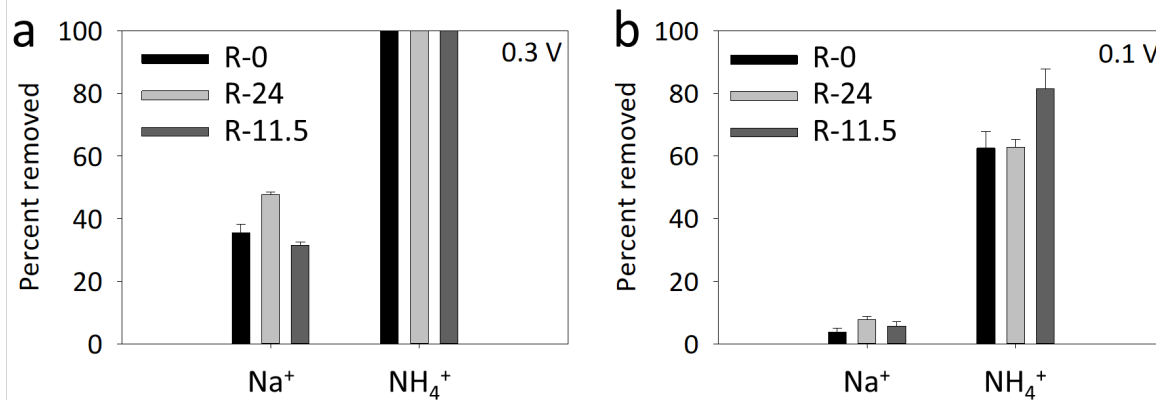


Fig. S2 Removal of cations (Na^+ and NH_4^+) depending on the chloride/phosphorus (Cl:P) ratio at the applied constant voltages of (a) 0.3 or (b) 0.1 V. NH_4^+ (5 mM) and Na^+ (20 mM) concentrations remained constant and the R-0, R-24, and R-11.5 solutions were used.

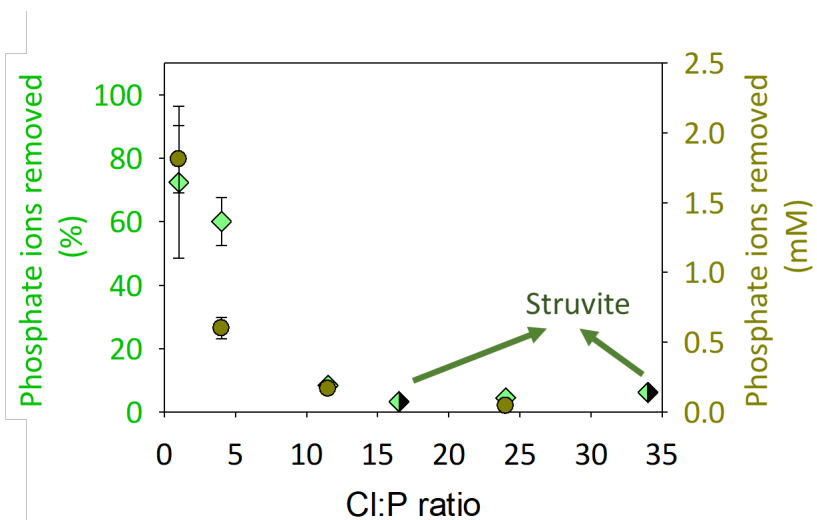


Fig. S3 Phosphate ions removal in terms of percent (trapezoid, green) and molar concentration (circle, dark yellow) as a function of chloride/phosphorus (Cl:P) ratio. In order to generate struvite ($\text{NH}_4\text{MgPO}_4\cdot 6\text{H}_2\text{O}$), 5 mM MgCl_2 was added to either R-11.5 or R-24 solution. The right-filled trapezoids indicate phosphate ions removal (%) obtained by the tests for struvite formation.